

Try It Out

1. $\text{MgO} + \text{Fe} \rightarrow \text{Fe}_2\text{O}_3 + \text{Mg}$
2. $\text{Cu}_2\text{S} + \text{O}_2 \rightarrow \text{Cu} + \text{SO}_2$
3. $\text{FeCl}_2 + \text{Na}_3(\text{PO}_4) \rightarrow \text{Fe}_3(\text{PO}_4)_2 + \text{NaCl}$
4. $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
5. $\text{Al} + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2\text{SO}_4 + \text{H}_2$
6. $\text{Zn} + \text{N}_2 \rightarrow \text{Zn}_3\text{N}_2$
7. $\text{K}_2\text{O} \rightarrow \text{K} + \text{O}_2$
8. $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
9. $\text{KClO}_3 + \text{Zn} \rightarrow \text{Zn}(\text{ClO}_3)_2 + \text{K}$
10. $\text{C}_4\text{H}_8\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
11. $\text{ZnCl}_2 + \text{K} \rightarrow \text{KCl} + \text{Zn}$
12. $\text{CH}_2\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$